

By saturating **II** with hydrogen bromide in an atmosphere of nitrogen first in the cold and then at 100° C we obtained 1-bromo-2-(β-bromopropyl)cyclohexane (**III**), the structure of which was confirmed by its IR spectrum. Bp 99–100° C (2 mm),  $n_D^{20}$  1.5280,  $d_4^{20}$  1.5251. Found, %: C 38.19; H 5.70; Br 56.34; MR<sub>D</sub> 57.09. Calculated for C<sub>9</sub>H<sub>16</sub>Br<sub>2</sub>, %: C 38.03; H 5.63; Br 56.23; MR<sub>D</sub> 57.03. Yield 86.3%.

The reaction of **III** with a threefold excess of anhydrous Na<sub>2</sub>S in dimethylformamide at 125° C gave a 54.5% yield of **I** in the form of a mixture of cis- and trans-isomers. Bp 89°–90° C (20 mm),  $n_D^{20}$  1.5090,  $d_4^{20}$  0.9783. Found, %: C 57.24; H 10.35; MR<sub>D</sub> 47.61. Cal-

culated for C<sub>9</sub>H<sub>16</sub>S, %: C 57.22; H 10.25; MR<sub>D</sub> 47.43. The structure of **I** was confirmed by its IR spectrum.

## REFERENCE

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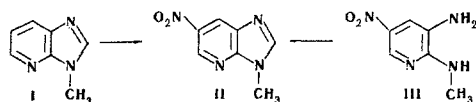
## NITRATION OF 3-METHYL-3H-IMIDAZO[4,5-b]PYRIDINE

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We have effected the direct introduction of a nitro group into the molecule of an imidazo[4,5-b]pyridine for the first time. In contrast to benzimidazole, the nitration of which takes place readily even in the cold [1], 3-methyl-3H-imidazo[4,5-b]pyridine (**I**) undergoes nitration with a mixture of nitric and sulfuric acids only at 140–160° C. The reaction product [yield 50%, mp 220–221° C (ethanol). Found, %: C 46.92; H 3.34; N 31.34. Calculated for C<sub>7</sub>H<sub>8</sub>N<sub>4</sub>O<sub>2</sub>, %: C 47.19; H 3.39; N 31.45] proved to be identical with the compound obtained from 3-amino-2-methylamino-5-nitropyridine (**III**) and formic acid, which shows its structure to be 3-methyl-6-nitroimidazo[4,5-b]-pyridine (**II**).



No other nitro derivatives apart from **II** were detected. Compound **III** was obtained by the reduction with ammonium sulfide of the product of the interaction of 2-chloro-3,5-dinitropyridine with methylamine [2]. Mp 199–200° C (water). Found, %: C 42.94; H 4.83. Calculated for C<sub>6</sub>H<sub>8</sub>N<sub>4</sub>O<sub>2</sub>, %: C 42.86; H 4.79.

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## QUATERNIZATION OF 3-METHYL-3H-IMIDAZO[4,5-b]PYRIDINE

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Khimiya Geterotsiklicheskikh Soedinenii, Vol. 4, No. 5, p. 954, 1968

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3-Methyl-3H-imidazo[4,5-b]pyridine (**I**), which has two basic centers [1], reacts with alkylating agents at the N<sup>1</sup> atom with the formation of monoquaternary salts. The structure of such compounds as N<sup>1</sup>-alkyl(aryl, aralkyl)-N<sup>3</sup>-methylimidazolium salts (**II**) has been shown by the oxidation of some of them (R = CH<sub>3</sub>, X = I; R = CH<sub>2</sub>C<sub>6</sub>H<sub>5</sub>, X = Cl) with potassium ferrocyanide in alkali at a temperature not exceeding 10° C to N<sup>1</sup>, N<sup>3</sup>-disubstituted imidazo[4,5-b]pyridin-2-ones (**III**) (R = CH<sub>3</sub>, CH<sub>2</sub>C<sub>6</sub>H<sub>5</sub>). The same substances have been obtained by the methylation with dimethyl sulfate and the benzylation with di-

methylphenylbenzylammonium hydroxide of 3-methylimidazo[4,5-b]pyridin-2-ones (**IV**) in an alkaline medium. Compound **IV** was synthesized by fusing 3-amino-2-methylaminopyridine with urea at 170° C.

